

	KS4 Science: Chemical Changes part 1	I can do this	Covered in Class	Strength ?	Revised it?	Kerboodle Textbook page reference
Review of previous test - work on targets (within first few lessons of new	<ul style="list-style-type: none"> I know where I could improve from the feedback I have understood what I need to do to improve I have acted on Feedback and edited in Blue Pen 					0
metal oxides and introduction to reactivity series (displacement	<ul style="list-style-type: none"> I can identify that metals react with oxygen to form metal oxides I can explain oxidation by gain of oxygen I can write equations to show transfer of electrons 					Oxford p84
Determining reactivity from reactions with water	<ul style="list-style-type: none"> I can describe the reactions of metals with water I can deduce the order of reactivity based on experimental results I can explain how the reactivity is related to the tendency of the metal to form its positive ion 					Oxford p84
Corrosion and its prevention (triple only)	<ul style="list-style-type: none"> I know how experimental results can be used to show the conditions for rusting I know how to protect iron from rusting I understand the term sacrificial protection 					Oxford p220-221
Extractions of metals with carbon	<ul style="list-style-type: none"> I can interpret processes used to extract different metals I can identify substances that are oxidised or reduced in terms of gain or loss of oxygen I can write half equations for the reactions 					Oxford p88-89
Ionic equations for displacement and redox	<ul style="list-style-type: none"> I can write half equations for a displacement reaction I know from a half equation which substance has been oxidised and reduced I understand what is happening in a displacement reaction in terms of movement of electrons 					Oxford p86-87
Reactions of acids and metals	<ul style="list-style-type: none"> I can describe how to make salts I can write full balanced symbol equations I can explain half equations in terms of oxidation and reduction 					Oxford p90-91
Review (Mid Topic Assessment)	<ul style="list-style-type: none"> 					0